REGIOSELECTIVE MANNICH REACTIONS VIA TRIMETHYLSILYL ENOL ETHERS

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<u>Summary</u>: Site-specific reactions of trimethylsilyl enolates and trimethylsilyldienolates are described.

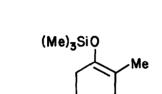
Recently some reactions of the Eschenmoser salt,  $l^1$ , with enolates<sup>2,3,4</sup>, Grignard<sup>5a</sup> and lithium<sup>5b</sup> reagents were described. This methodology was used to some advantage in a synthesis of vernolepin.<sup>6</sup> A more detailed study was provided by Holy.<sup>7</sup>

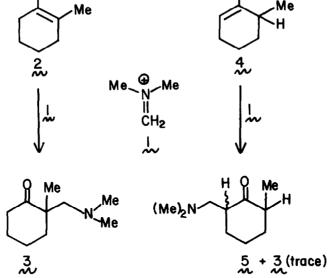
In our preliminary experiments<sup>2</sup> we also noted that compound 1 is sufficiently electrophilic to react with trimethylsilyl enol ethers. This process was used in the synthesis of a known precursor of dl-progesterone. We now describe a fuller examination of the reactions of 1 with trimethylsilyl enol ethers. In particular we were attracted by the possibility of exploiting such methodology to address the longstanding problem of regiospecificity in the Mannich Reaction.<sup>8a</sup> While some solutions to this problem have been advanced in the literature<sup>8b</sup>, the possibility of obtaining positionally defined Mannich bases via positionally defined silyl enol ethers<sup>9</sup> seemed most attractive.

Treatment of 1-trimethylsilyloxy-2-methylcyclohexene (2) with the iodide, 1, at room temperature for 45 min followed by acidic workup and basification afforded a 65% yield of a Mannich base. Singlet resonances for the quaternary bound methyl group  $\delta$ 1.1 (s, 3) and for the NCH<sub>2</sub> protons  $\delta$ 2.54 (s, 2) served to define the structure of the Mannich base as 3.10

Similarly, reaction of  $\frac{4}{2}$  with  $\frac{1}{2}$  afforded a 79% yield of  $\frac{5}{2}$  as a 2:1 mixture of stereosiomers. With both  $\frac{3}{2}$  and  $\frac{5}{2}$  in hand, it was possible to probe the regiospecificity of the reactions of  $\frac{2}{2}$  and  $\frac{4}{2}$  with  $\frac{1}{2}$  in greater detail. Within the limits of nmr detection at 250 MHz, we could find no indication that any component of isomer mixture  $\frac{5}{2}$  had risen in the reaction of  $\frac{1}{2} + \frac{2}{2}$ . However, there could be detected <u>ca</u> 4% of  $\frac{3}{2}$  in the reaction of  $\frac{1}{2} + \frac{4}{2}$ . Since starting silyl enol ether  $\frac{4}{2}$  was free of detectable amounts of its isomer,  $\frac{2}{2}$  there appears to be a very slight loss of specificity in the latter case.

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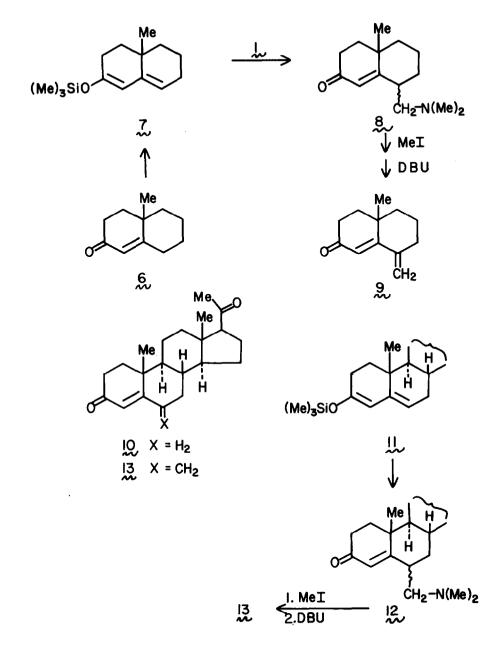


(Me)<sub>3</sub>SiO

We next investigated the reactions of conjugated silvldienolates<sup>11</sup> with compound 1. The hope, that such a process would provide a means for the introduction of carbon functionality at the  $\gamma$ -position of an enone, was realized in practice. Reaction of siloxydiene  $\chi$  (prepared from the readily available  $\delta$  in the usual way) with 1 (chloroform; reflux; 30 min) afforded a 65% yield of  $\delta$  (single epimer; configuration at C<sub>6</sub> not rigorously determined). Conversion of  $\delta$  to the  $\gamma$ -methyleneketone,  $9^{10}$ , was accomplished in 73% yield as shown. The overall method for introduction of a  $\gamma$ -methylene group onto  $\delta$  appears to be much simpler than that recently described by Yoshikoski.<sup>12</sup>

The process was readily extended to the steroid series. Thus, without protection of the C<sub>20</sub> ketone, progesterone (10) was converted to the silyldienolate  $11^{10}$ , mp 80-81°. Reaction of 11 with 1, as above, afforded oily 12 (apparently a single isomer, stereochemistry at C<sub>6</sub> not determined), which was converted to  $13^{10}$ , mp 105-106°, in 58% yield in the same fashion.

In summary, it would appear that the reactions of silylenolates and silyldienolates with 1 provide excellent regiospecificity for Mannich reactions. The use of this methodology in the context of larger synthetic programs is a matter under investigation.



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